

FOUR PHENYLETHANOIDS FROM LEAVES OF *PSEUDERANTHEMUM CARRUTHERSII* (SEEM.) GUILL. VAR. *ATROPURPUREUM* (BULL.) FOSB.

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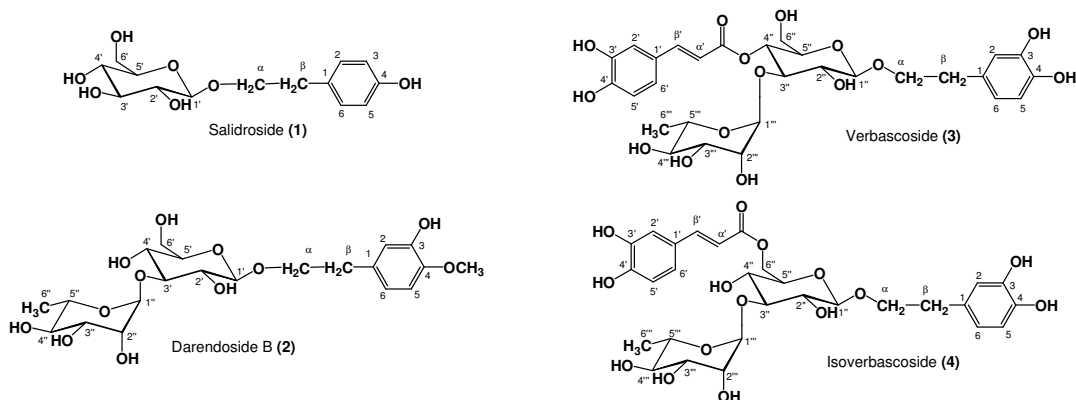
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ABSTRACT

Four phenylethanoids were isolated from the dried leaves of *Pseuderanthemum carruthersii* (Seem.) Guill. var. *atropurpureum* (Bull.) Fosb. (Acanthaceae), including *solidroside* (1), *darendoside B* (2), *verbascoside* (3) and *isoverbascoside* (4). Their chemical structures were elucidated by spectroscopic methods. These substances were isolated for the first time from this genus.



I - INTRODUCTION

P. carruthersii (Seem.) Guill. var. *atropurpureum* (Bull.) Fosb. (Acanthaceae) has been used in traditional medicine to heal the wound [1]. It has not yet much been studied. In this paper, from the leaves of this plant, four phenylethanoids were isolated and their chemical structures were elucidated by spectroscopic means.

II - EXPERIMENTAL

General experimental procedures

The NMR spectra were measured on a Bruker Avance spectrometer, at 500 MHz for ¹H and 125 MHz for ¹³C, in the Institute of Chemistry, Vietnamese Academy of Science and Technology. The HR-ESI-MS were recorded on a HR-ESI-MS microOTOF-Q

10187 mass spectrometer in the University of Science, National University – HCM City.

Plant material

The leaves of *Pseuderanthemum carruthersii* (Seem.) Guill. var. *atropurpureum* (Bull.) Fosb., collected in June 2008 in Binh Phuoc province, were identified by Dr. Hoang Viet, Department of Biology, University of Science, National University – HCM City. A voucher specimen (US–A007) was deposited in the herbarium of the Department of Organic Chemistry, University of Science, National University – HCM City.

Extraction and Isolation

The dried leaves of *P. carruthersii atropurpureum* (5 kg) were exhaustively extracted with methanol by maceration at room temperature and the methanolic filtrate was concentrated *in vacuo* to yield a residue of 800g. This crude residue was suspended in water and partitioned between water and petroleum ether and then ethyl acetate. The remaining aqueous solution was subjected to Diaion HP–20 column chromatography, and eluted with water, 50% methanol and methanol, respectively. The 50% methanol residue (30g) was applied on a RP–18 silica gel chromatographic column eluted with water–methanol (8:1 to 2:1) to give 7 fractions. Fraction 4 (5.3g) was purified by silica gel column chromatography eluted with ethyl acetate: methanol: water 8:1:1 to afford compound **1** (51mg). Fraction 5 (3.3g) was purified by silica gel column chromatography eluted with ethyl acetate: methanol: water 8:1:1 to afford compound **2** (35mg). Fraction 6 (3.7g) was applied on a silica gel chromatographic column with chloroform:methanol:water 20:6:1 to afford compound **3** (42mg) and compound **4** (38mg).

* **Salidroside (1)**: amorphous powder; ^1H and ^{13}C NMR (CD_3OD) see Table 1, HR-ESI-MS m/z 323.1120 $[\text{M}+\text{Na}]^+$.

* **Darendoside B (2)**: amorphous powder; ^1H and ^{13}C NMR (CD_3OD) see table 1, HR-ESI-

MS m/z 499.1742 $[\text{M}+\text{Na}]^+$.

* **Verbascoside (3)**: amorphous powder; ^1H and ^{13}C NMR (CD_3OD) see Table 2, HR-ESI-MS m/z 647.1972 $[\text{M}+\text{Na}]^+$.

* Isoverbascoside (4): amorphous powder; ^1H and ^{13}C NMR (CD_3OD) see table 2, HR-ESI-MS m/z 647.1985 $[\text{M}+\text{Na}]^+$.

III - RESULTS AND DISCUSSION

The methanolic extract of the dried leaves of *Pseuderanthemum carruthersii* (Seem.) Guill. var. *atropurpureum* (Bull.) Fosb. was separated by silica gel column chromatography to afford four compounds, **1–4**. Compound **1** was isolated as amorphous powder, ESI-MS showed m/z 323.1120 $[\text{M}+\text{Na}]^+$ supported the molecular formula $\text{C}_{14}\text{H}_{20}\text{O}_7$. The ^1H spectrum (table 1) of **1** showed the presence of aromatic protons at δ_{H} 7.08 (2H, *d*, $J = 8.5\text{Hz}$, H–2 and H–6) and 6.72 (2H, *d*, $J = 8.5\text{ Hz}$, H–3 and H–5) assigned to a *p*-substituted benzene ring. The substituent located at C–4 should be a hydroxyl group because the chemical shift value of C–4 moved to the low field at δ_{C} 156.8. The COSY experiment exhibited the correlation of δ_{H} 3.73 (1H, *m*, H– α) to δ_{H} 2.85 (2H, *dt*, $J = 3.0, 8.0\text{ Hz}$, H– β) proving C– α next to C– β . The HMBC correlation of H– β at δ_{H} 2.85 to C–1 at δ_{C} 130.8 showed the attachment of C– β to the aromatic ring. The above information demonstrated that **1** has a skeleton of phenylethanoid. The β –glucopyranosyl moiety was demonstrated by the presence of the anomeric proton at δ_{H} 4.31 (1H, *d*, $J = 8.0\text{ Hz}$, H–1') corresponding to carbon C–1' at δ_{C} 104.4 as well as the doublet of doublet at δ_{H} 3.88 (1H, *dd*, $J = 12.0, 2.0\text{ Hz}$, H–6') and δ_{H} 3.69 (1H, *dd*, $J = 12.0, 5.0\text{ Hz}$, H–6') corresponding to C–6' at δ_{C} 62.8. Additionally, ^1H NMR showed signals from 3.00 - 4.00 ppm corresponding to carbon signals from 60-80 ppm for carbons linking to hydroxyl groups. The HMBC correlation from the proton H–1' (δ_{H} 4.31) to C– α (δ_{C} 72.1) showed the attachment of the glucopyranosyl moiety at C– α of phenylethyl

moiety. By this information and the comparison of the published data ^[2], the compound **1** was identified as salidroside, known as β -(4-hydroxyphenyl)ethyl-*O*- β -D-glucopyranoside.

Table 1: 1D and 2D - NMR spectral data of **1** and **2** (CD₃OD)

No	Compound 1 (CD ₃ OD)			Salidroside (DMSO- d ₆) ^[2]	Compound 2 (CD ₃ OD)			Darendoside B (CD ₃ OD) ^[4]
	δ_H (ppm) <i>J</i> (Hz)	δ_C (ppm)	HMBC (¹ H→ ¹³ C)	δ_C (ppm)	δ_H (ppm) <i>J</i> (Hz)	δ_C (ppm)	HMBC (¹ H→ ¹³ C)	δ_C (ppm)
1	-	130.8	-	128.9	-	132.9	-	132.8
2	7.08 (<i>d</i> , 8.5)	130.9	1, 3, 4, 6, β	130.1	6.75 (<i>d</i> , 2.0)	117.1	4, 6	112.8 (**)
3	6.72 (<i>d</i> , 8.5)	116.1	1, 2, 4, 5	115.3	-	147.5	-	147.4
4	-	156.8	-	155.9	-	147.4	-	147.2
5	6.72 (<i>d</i> , 8.5)	116.1	1, 3, 4, 6	115.3	6.83 (<i>d</i> , 8.5)	112.9	1, 3, 6	117.0 (**)
6	7.08 (<i>d</i> , 8.5)	130.9	1, 2, 4, 5, β	130.1	6.70 (<i>dd</i> , 8.5, 2.0)	121.2	2, 4	121.1
OCH ₃					3.83(<i>s</i>)	56.5	3	56.5
α	4.05 (<i>m</i>) 3.73 (<i>m</i>)	72.1	1, 1', β	70.2	4.06 (<i>m</i>) 3.73 (<i>m</i>)	72.3	1, 1'	71.9
β	2.85 (<i>dt</i> , 3.0, 8.0)	36.4	1, α	35.1	2.82 (<i>dt</i> , 3.0, 7.5)	36.6	1, 2, 6, α	36.5
1'	4.31 (<i>d</i> , 8.0)	104.4	3', α	103.1	4.31 (<i>d</i> , 8.0)	104.2	α	104.1
2'	3.20 (<i>dd</i> , 9.0, 8.0)	75.1	1', 3'	73.7	3.28 (*)	75.6	1', 3'	75.5
3'	3.37 (<i>m</i>)	78.1	2', 4'	77.1	3.51 (<i>t</i> , 9.0)	84.5	2', 4'	84.4
4'	3.25 – 3.55	71.7	5'	70.4	3.37 (<i>t</i> , 9.0)	70.2	5', 6'	70.1
5'	3.25 – 3.55	77.9	4', 6'	77.2	3.33 (*)	77.9	3', 6'	77.7
6'a	3.88 (<i>dd</i> , 12.0, 2.0)	62.8	4', 5'	61.4	3.88 (<i>dd</i> , 12.0, 2.0)	62.7		62.6
6'b	3.69 (<i>dd</i> , 12.0, 5.0)				3.69(*)			
1''					5.17 (<i>d</i> , 1.5)	102.8	2'', 5'', 3'	102.6
2''					3.96 (<i>dd</i> , 3.0, 1.5)	72.4	4''	72.2
3''					3.71 (*)	72.0	4''	72.1
4''					3.42 (<i>t</i> , 9.5)	74.0	3'', 5''	73.9
5''					4.01 (*)	70.1		70.0
6''					1.27 (<i>d</i> , 6.0)	17.9		17.9

(*) Crowded signal, difficult to be analysed

(**) Interchangeable

Compound **2** was isolated as amorphous powder, ESI-MS showed *m/z* 499.1742 [M+Na]⁺ supported the molecular formula C₂₁H₃₂O₁₂. The ¹H spectrum (table 1) of **2**

revealed the presence of an ABX system [δ_{H} 6.83 (1H, *d*, 8.5, H-5), 6.75 (1H, *d*, 2.0, H-2) and 6.70 (1H, *dd*, 8.5, 2.0, H-6)] for a 1,3,4-trisubstituted benzene ring. The ^1H spectrum showed the presence of a methoxyl group because of δ_{H} 3.83 (3H, *s*). The HMBC correlation of H of methoxyl group (δ_{H} 3.83) to C-4 (δ_{C} 147.4) provided that the methoxyl group located at C-4. The chemical shift value C-3 moved to the low field at δ_{C} 147.5, so that the substituent at C-3 should be hydroxyl group. The COSY experiment exhibited the correlation of δ_{H} 3.73 (1H, *m*, H- α) to δ_{H} 2.82 (2H, *dt*, $J = 3.0, 7.5$ Hz, H- β) proving C- α next to C- β . The HMBC correlation of H- β at δ_{H} 2.82 to C-1 at δ_{C} 132.9 showed the attachment of C- β to the aromatic ring. The above information demonstrated that **2** is a skeleton of phenylethanoid. The β -glucopyranosyl moiety was demonstrated by the presence of the anomeric proton at δ_{H} 4.31 (1H, *d*, $J = 8.0$ Hz, H-1') corresponding to carbon C-1' at δ_{C} 104.2 as well as a doublet of doublet at δ_{H} 3.88 (1H, *dd*, $J = 12.0, 2.0$ Hz, H-6'a) and δ_{H} 3.69 (1H, *m*, H-6'b) corresponding to C-6' at δ_{C} 62.7. The HMBC correlation from the proton H-1' (δ_{H} 4.31) to C- α (δ_{C} 72.3) showed the attachment of the glucopyranosyl moiety at C- α of phenylethyl moiety. The α -rhamnopyranosyl moiety was demonstrated by the presence of the anomeric proton at δ_{H} 5.17 (1H, *d*, $J = 1.5$ Hz, H-1'') corresponding to carbon C-1'' at δ_{C} 102.8 as well as a doublet at δ_{H} 1.27 (3H, *d*, $J = 6.0$ Hz, H-6'') corresponding to C-6'' at δ_{C} 17.9. The HMBC correlation from the proton H-1'' (δ_{H} 5.17) to C-3' (δ_{C} 84.5) showed the attachment of the rhamnopyranosyl moiety at C-3' of glucopyranosyl moiety. Based on this information and the comparison of the published data [3, 4], the compound **2** was identified as darendoside B, known as [β -(3-hydroxy-4-methoxyphenyl)ethyl]-(3'-*O*- α -L-rhamnopyranosyl)-*O*- β -D-glucopyranoside.

Compound **3** was isolated as amorphous

powder, ESI-MS showed m/z 647.1972 $[\text{M}+\text{Na}]^+$ supported the molecular formula $\text{C}_{29}\text{H}_{36}\text{O}_{15}$. The ^1H spectrum (Table 2) of **3** revealed the presence of an ABX system [δ_{H} 6.71 (1H, *d*, 8.0, H-5), 6.72 (1H, *d*, 2.0, H-2) and 6.59 (1H, *dd*, 8.0, 2.0, H-6)] for a 1,3,4-trisubstituted benzene ring. The chemical shift values of C-3 and C-4 moved to the low field at δ_{C} 146.1 and 144.6, respectively, so that the substituents at C-3 and C-4 should be hydroxyl groups. The COSY experiment exhibited the correlation of δ_{H} 3.75 (1H, *m*, H- α) to δ_{H} 2.81 (2H, *dt*, $J = 3.0, 7.5$ Hz, H- β) proving C- α next to C- β . The HMBC correlation of H- β at δ_{H} 2.81 to C-1 at δ_{C} 131.5 showed the attachment of C- β to the aromatic ring. The above information demonstrated that **3** has a skeleton of 3,4-dihydroxyphenylethyl moiety. The ^1H spectrum showed the presence of another ABX system [δ_{H} 7.07 (1H, *d*, 2.0, H-2'), 6.98 (1H, *dd*, 8.0, 2.0, H-6') and 6.80 (1H, *d*, 8.5, H-5')] for a 1,3,4-trisubstituted benzene ring and two *trans*-olefinic protons [δ_{H} 7.62 (1H, *d*, 16.0, H- β') and 6.29 (1H, *d*, 16.0, H- α')]. The chemical shift values of C-3' and C-4' moved to the low field at δ_{C} 146.8 and 149.1, respectively, so that the substituents at C-3' and C-4' should be hydroxyl groups. The HMBC correlation of H- β' at δ_{H} 7.62 to C-2' at δ_{C} 115.3, C-6' at δ_{C} 123.2 and $>\text{C}=\text{O}$ at δ_{C} 168.3 proved the caffeoyl moiety. The β -glucopyranosyl moiety was demonstrated by the presence of the anomeric proton at δ_{H} 4.39 (1H, *d*, $J=8.0$ Hz, H-1''') corresponding to carbon C-1''' at δ_{C} 104.2. The HMBC correlation from the proton H-1''' (δ_{H} 4.39) to C- α (δ_{C} 72.2) showed the attachment of the glucopyranosyl moiety at C- α of phenylethyl moiety. The rhamnopyranosyl moiety was demonstrated by the presence of the anomeric proton at δ_{H} 5.21 (1H, *d*, $J = 1.0$ Hz, H-1''') corresponding to carbon C-1'''' at δ_{C} 103.0 as well as a doublet at δ_{H} 1.11 (3H, *d*, $J = 6.0$ Hz, H-6''') corresponding to C-6'''' at δ_{C} 18.4. The HMBC correlation from the proton H-1'''' (δ_{H} 5.21) to C-3''' (δ_{C} 81.6) showed the attachment of the

rhamnopyranosyl moiety at C-3'' of glucopyranosyl moiety. The HMBC correlation from H-4'' (δ_{H} 4.94) of glucopyranosyl moiety to $>\text{C}=\text{O}$ (δ_{C} 168.3) indicated the location of the caffeoyl moiety to glucopyranosyl moiety at C-4''. Based on this information and the comparison of the published data [5], the compound (**3**) was identified as verbascoside, known as acteoside or [β -(3,4-dihydroxyphenyl)ethyl]-(3''-*O*- α -L-rhamnopyranosyl)-(4''-*O*-caffeoyl)-*O*- β -D-glucopyranoside.

Compound **4** was isolated as amorphous powder, ESI-MS showed m/z 647.1985 [$\text{M}+\text{Na}$]⁺ supported the molecular formula $\text{C}_{29}\text{H}_{36}\text{O}_{15}$. The NMR spectra of compound **4** were similar to those of compound **3**. The ¹H spectrum (Table 2) of **4** revealed the presence of two sets of ABX systems [δ_{H} 6.69 (1H, *d*, 2.0, H-2), 6.66 (1H, *d*, 8.0, H-5) and 6.55 (1H, *dd*, 8.0, 2.0, H-6)] for the 3,4-dihydroxyphenylethyl moiety and [δ_{H} 7.05 (1H, *d*, 2.0, H-2'), 6.91 (1H, *dd*, 8.0, 2.0, H-6') and 6.80 (1H, *d*, 8.0, H-5')] for the caffeoyl moiety,

two *trans*-olefinic protons [δ_{H} 7.58 (1H, *d*, 16.0, H- β') and 6.30 (1H, *d*, 16.0, H- α')], together with two anomeric protons at δ_{H} 4.35 (1H, *d*, 8.0, H-1'') for β -D-glucose and δ_{H} 5.20 (1H, *d*, 1.5, H-1''') for α -L-rhamnose. In the HMBC experiment, the correlations of H-1'' (δ_{H} 4.35) to C- α (δ_{C} 72.4), H-1''' (δ_{H} 5.20) to C-3'' (δ_{C} 84.0), together with chemical shift values of these protons and carbons revealed the connection among the 3,4-hydroxyphenylethyl and the two sugar moieties. So **4** is a positional isomer of **3** with the differently positional connectivity between the caffeoyl and the glucopyranosyl moieties. The HMBC correlation from H-6'' (δ_{H} 4.52) to $>\text{C}=\text{O}$ (δ_{C} 169.1) indicated the location of the caffeoyl moiety to the glucopyranosyl moiety at C-6''. Based on this information and the comparison of the published data [6, 7], the compound **4** was identified as isoverbascoside, known as isoacteoside or [β -(3,4-dihydroxyphenyl)ethyl]-(3''-*O*- α -L-rhamnopyranosyl)-(6''-*O*-caffeoyl)-*O*- β -D-glucopyranoside.

Table 2: 1D and 2D - NMR spectral data of **3** and **4** (CD₃OD)

No	Compound 3 (CD ₃ OD)			Verbascoside (CD ₃ OD) ^[5]	Compound 4 (CD ₃ OD)			Isoverbascoside (CD ₃ OD) ^[7]
	δ_{H} (ppm) <i>J</i> (Hz)	δ_{C} (ppm)	HMBC (¹ H→ ¹³ C)	δ_{C} (ppm)	δ_{H} (ppm) <i>J</i> (Hz)	δ_{C} (ppm)	HMBC (¹ H→ ¹³ C)	δ_{C} (ppm)
(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)
1	-	131.5	-	131.6	-	131.4	-	131.4
2	6.72 (<i>d</i> , 2.0)	117.1	4, 6	117.3	6.69 (<i>d</i> , 2.0)	117.0	3, 6	117.0
3	-	146.1	-	146.9	-	146.1	-	144.5 (**)
4	-	144.6	-	144.8	-	144.6	-	146.0 (**)
5	6.71 (<i>d</i> , 8.0)	116.3	1, 3	116.6	6.66 (<i>d</i> , 8.0)	116.3	1, 4	116.3
6	6.59 (<i>dd</i> , 8.0, 2.0)	121.3	2, 4	121.4	6.55 (<i>dd</i> , 8.0, 2.0)	121.3	2, 3	121.3
α	4.07 (<i>m</i>) 3.75 (<i>m</i>)	72.2	1, 1'' -	72.2	3.91 (<i>m</i>) 3.71- 3.76 (*)	72.4	1, 1', β	72.4
β	2.81 (<i>dt</i> , 3.0, 7.5)	36.5	α , 1, 2, 6	36.7	2.79 (<i>dt</i> , 1.5, 6.5)	36.6	1, 2, 6, α	36.5
1'	-	127.7	-	127.8	-	127.7	-	127.6

(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)
2'	7.07 (<i>d</i> , 2.0)	115.3	3', 4', 6'	115.4	7.05 (<i>d</i> , 2.0)	115.1	1', 3', 6'	115.1
3'	-	146.8	-	146.2	-	146.7	-	149.5 (**)
4'	-	149.1	-	149.9	-	149.5	-	146.6 (**)
5'	6.80 (<i>d</i> , 8.5)	116.5	1', 3', 4'	116.5	6.80 (<i>d</i> , 8.0)	116.5	1', 3', 4'	116.5
6'	6.98 (<i>dd</i> , 8.0, 2.0)	123.2	β', 2', 4'	123.4	6.91 (<i>dd</i> , 8.0, 2.0)	123.1	2', 3', β'	123.1
α'	6.29 (<i>d</i> , 16.0)	114.7	1', C=O	114.8	6.30 (<i>d</i> , 16.0)	114.8	1', C=O	114.8
β'	7.62 (<i>d</i> , 16.0)	148.0	α', 2', 6', C=O	148.2	7.58 (<i>d</i> , 16.0)	147.2	α', 6', C=O	147.2
C=O	-	168.3	-	168.5	-	169.1	-	169.1
1''	4.39 (<i>d</i> , 8.0)	104.2	α	104.3	4.35 (<i>d</i> , 8.0)	104.3	α	104.2
2''	3.41 (<i>dd</i> , 9.0, 8.0)	76.2	1'', 3''	76.3	3.33-3.40 (*)	75.4	1'', 3''	75.2
3''	3.83 (<i>t</i> , 9.0)	81.6	1''', 2'', 4''	81.8	3.52-3.59 (*)	84.0	2'', 4'', 1'''	83.9
4''	4.94 (<i>t</i> , 9.5)	70.4	3'', 5'', C=O	70.5	3.40-3.50 (*)	70.0	5'', 6''	70.0
5''	3.50-3.66 (*)	76.0	1''	76.1	3.52-3.59 (*)	75.6		75.5
6''a	3.50-3.66 (*)	62.4	4''	62.5	4.52 (<i>dd</i> , 12.0, 2.0)	64.6	C=O	64.6
6''b	3.50-3.66 (*)				4.38 (<i>dd</i> , 12.0, 6.0)			
1'''	5.21 (<i>d</i> , 1.0)	103.0	3'', 2''', 5'''	103.2	5.20 (<i>d</i> , 1.5)	102.7	3'', 2''', 5'''	102.6
2'''	3.94 (<i>m</i>)	72.3	3''', 4'''	72.5	3.95-4.04 (*)	72.2	3'''	72.2
3'''	3.50-3.66	72.0	4'''	72.4	3.71-3.76	72.3	4'''	72.2
4'''	3.31 (<i>m</i>)	73.8	3''', 5'''	73.9	3.40-3.50	74.0	3''', 5'''	73.9
5'''	3.50-3.66	70.6		70.7	3.95-4.04	70.4	4'''	70.3
6'''	1.11 (<i>d</i> , 6.0)	18.4	4''', 5'''	18.6	1.29 (<i>d</i> , 6.0)	17.8	4''', 5'''	17.8
(*) Crowded signal, difficult to be analysed						(**) Interchangeable		

IV - CONCLUSION

From the leaves of *Pseuderanthemum carruthersii* (Seem.) Guill. Var. *atropurpureum* (Bull.) Fosb. collected in June 2008 in Binh Phuoc province, four phenylethanoids were isolated and elucidated as salidroside (**1**), darendoside B (**2**), verbascoside (**3**) and isoverbascoside (**4**). These substances were isolated for the first time from this genus. Further studies are conducting on this plant.

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**BỐN PHENYLETHANOID ĐƯỢC CÔ LẬP TỪ CÂY XUÂN HOA ĐỎ,
PSEUDERANTHEMUM CARRUTHERSII (SEEM.) GUILL. VAR. *ATROPURPUREUM*
(BULL.) FOSB.**

Tóm tắt:

Từ bột khô lá cây Xuân hoa đỏ, *Pseuderanthemum carruthersii* (Seem.) Guill. var. *atropurpureum* (Bull.) Fosb., họ Ô rô (Acanthaceae), đã cô lập được bốn phenylethanoid, đó là: salidroside (1), darendoside B (2), verbascoside (3) and isoverbascoside (4). Cấu trúc của các hợp chất này được xác định bằng các phương pháp phổ nghiệm và so sánh với tài liệu tham khảo. Các hợp chất này được cô lập lần đầu tiên trong chi *Pseuderanthemum*.