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Study on solidified material from dredged sediment, fly ash, and blended portland cement using the response surface method

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Abstract. Treating dredged sediment is a complex processing and ongoing challenge. To utilize dredged sediment for the landfill or construction purposes, a material fabricated from a mixture of dredged sediment, Portland cement, and fly ash, was cured under room temperature and hydrothermal condition at 180 °C and 0.9 MPa pressure for 16 hours. The response surface methodology was used to evaluate the compressive strength of the material, with the range of factors investigated being the dredged sediments/solid ratio (0.3 - 0.9), cement/fly ash ratio (2 - 4), and water/solid ratio (0.45 - 0.55). The fitting models offered an accurate and reliable match to the actual data. The optimum mix proportions of two curing conditions were obtained using total desirability function, meet multi-objective criteria. This result finger out hydrothermal curing significantly enhances treatment capacity of dredged sediment, with a lower CO₂ emission in the mixture compared to ambient curing. Scanning electron microscopy (SEM), X-ray diffraction (XRD), and Fourier transform infrared spectroscopy (FTIR) were used to figure out the difference between the minerals formed in the material under two curing conditions, such as tobermorite.

Keywords: dredged sediment, response surface methodology, solidification, hydrothermal, tobermorite, multi-objective optimization.

Classification numbers: 3,4.3, 2.9.4, 2.10.2

1. INTRODUCTION

Dredged sediments (DS) include a diverse range of materials from marine dredging operations, continental watercourses, and the erosion of land masses. They consist of fine particles resulting from sediment deposition, introducing particles into ecosystems, and the precipitation of substances through biochemical processes in aquatic environments [1].

The growing volume of dredged sediments, estimated at approximately 600 Mm³ annually worldwide [2], presents several challenges, particularly considering their potential pollution and detrimental environmental effects. Consequently, there is a pressing need to identify optimal management approaches for these sediments.

Conventional approaches to managing these sediments involve landfilling and confined or unconfined disposal in the ocean. Nonetheless, these methods have numerous drawbacks, including high costs, limited capacity, and the potential for environmental contamination [3, 4]. Given the substantial quantities of sediments, alternative solutions should be explored to dispose of dredged materials properly.

The application of cementitious materials for stabilization/solidification (S/S) is a wellestablished and effective method for enhancing the engineering properties of sediments while encapsulating contaminants [5 - 7]. The process involves incorporating chemical compounds into the dredged material, aiming to achieve two objectives [8]:

- (i) chemically immobilizing the contaminants to decrease their leachability and bioavailability,
- (ii) geomechanically stabilizing the material to enable its reuse as new construction material.

Portland cement is widely preferred for stabilization/solidification due to its mechanical properties, widespread availability, and cost-effectiveness. To reduce environmental effects and improve mechanical properties, Portland cement (PC) can be partially replaced with pozzolanic materials like fly ash (FA) and ground granulated blast furnace slag (GGBS)...[9, 10]. Stabilized/solidified sediment is reused as a construction material instead of depleting other natural resources, which could be used as fill material [11] or bricks [12].

The compressive strength is widely employed as a parameter for assessing the mechanical strength of construction and road materials. In the case of using stabilized/solidified material as a low-strength filling material, compressive strength should exceed 1.15 MPa at 28 days of maintenance [13]. According to ASTM C62-10 standard, the minimum compressive strength requirement for building bricks is 10.3 MPa, with negligible weathering. Therefore, the compressive strength of treated sediments is the criterion used in the current study.

The presence of several components in mixed materials causes variation in their physical and chemical properties due to the coexistence of cement hydration and mineral mixtures, which affect the kinetics of hydration processes and the mechanisms of microstructure formation, the consequences of which are compressive strength [14]. These effects are further influenced by different curing conditions [15].

The studies conducted by Ribeiro et al. [16] and Consoli et al. [17], figure out that both the cement dosage and water content play a crucial role in determining the compressive strength values. In addition, the pozzolanic characteristic of FA makes a significant contribution to the improved mechanical performance proven by the compressive strength test with 9 % FA in a mixture of sediments and cement [18]. Further, adding 10 % FA to a mixture of sediments and cement was found to be the most effective in terms of strength [19, 20].

In order to optimize experimental conditions, the Response Surface Methodology is used. It is the optimal solution for giving the response function of many influencing factors at different levels without considering the physical nature of the process. The RSM enables the establishment of functional relationships between influential factors and corresponding response values. By fitting regression equations and visualizing response surfaces and contours, it is easier to identify the response values for each factor level [21]. Additionally, the RSM helps find the best predictors based on the response values at each factor level. Compared to single-factor control methods and orthogonal testing, RSM offers significant advantages in optimization design. In recent years, RSM has also been used to determine the mixing ratio of mortar and concrete [22 - 24].

In this study, the proportion of the mixture's factors affecting the compressive strength at different curing conditions (hydrothermal and room temperature curing) of solidified materials such as PC, DS, and FA were conducted according to the experimental Box-Behnken Design. Response functions are regression equations statistically processed and presented in a graphical form called response surfaces. In this study, the regression equations were processed to determine the proportion of DS in the mixture with the maximum proportion and minimum PC proportion to achieve the compressive strength threshold of concrete brick after 7 days (7.5 MPa) of curing under different conditions.

2. MATERIALS AND METHODS

2.1. Materials

Freshly dredged sediment was gathered from Cai Mep-Thi Vai International Port in Ba Ria - Vung Tau Province, Viet Nam. Once in the laboratory, the soil was dried under 110 °C for two days. After that, the soil was crushed, ground to under 200 μm, and homogenized.

Cement is a blended Portland cement with brand INSEE, type PCB40, conforming to Vietnamese National Standards TCVN 2682:2020. Clinker content is between 52 to 57 %, limestone is the main natural component of this cement.

Fly ash was obtained from Duyen Hai 1 Thermal Power Plant, located in Tra Vinh Province, Viet Nam.

2.2. Experimental Methods

2.2.1. Response surface methodology (RSM) and statistical analysis

Independent verichles	Coded	Levels			
independent variables	symbol	-1	0	1	
DS/(DS+PC+FA) ratio	x ₁	0.3	0.6	0.9	
PC/FA ratio	x ₂	2	3	4	
W/(DS+PC+FA) ratio	X ₃	0.4	0.45	0.5	

Table 1. Factors and levels in Box-Behnken design.

According to some previous studies [18, 25 - 27], the important independent variables affecting the compressive strength of solidified soil and dredged sediment are the type of binder, binder and additive content, soil properties, water content, curing time... However, this study focused on the influence of the DS, PC, FA, and water (W) mixing ratio on compressive strength of solidified dredged sediments by the response surface method with the Box-Behnken design. To simplify the number of variables, the ratio DS/(DS+PC+FA), ratio PC/FA, and ratio

W/(DS+PC+FA) were selected as independent variables. The two groups of response values correspond to the compressive strength of the curing material at room temperature (y_1) and hydrothermal condition (y_2) at 7 days (7d). The coding and non-coding of the three grade variants are presented in Table 1. A total of 15 experimental runs were conducted, with three replicates at the central points.

The response surface mix design is presented in Table 2. The models were fitted using regression analysis and analysis of variance (ANOVA), as well as the statistical significance of the quadratic model. The relationship between the response and each variable was visualized through three-dimensional (3D) surface and contour plots using Design Expert 13.0.5.0 software.

Run	x ₁	x ₂	X3
1	0.3	2	0.45
2	0.9	2	0.45
3	0.3	4	0.45
4	0.9	4	0.45
5	0.3	3	0.4
6	0.9	3	0.4
7	0.3	3	0.5
8	0.9	3	0.5
9	0.6	2	0.4
10	0.6	4	0.4
11	0.6	2	0.5
12	0.6	4	0.5
13	0.6	3	0.45
14	0.6	3	0.45
15	0.6	3	0.45

Table 2	Dasign	ofor	norimo	ntal
Tuble 2.	Design	OI EX	perme	mai

2.2.2. Specimen preparation

To obtain a homogeneous mixture, the 100 g of DS, PC, and FA were initially mixed for 1 minute, followed by adding water and an additional mixing period of 4 minutes. The paste mixture was then poured into a cylindrical PVC mold (21 mm diameter x 42 mm height) and shaken for 2 minutes to remove air voids. To ensure the reliability of the experimental data, three identical samples were prepared for each group. After 48 hours, the samples were disassembled, divided into two groups. The first group is further cured until the desired curing time is reached. The second group is cured under hydrothermal condition (temperature 180 °C, pressure 0.9 MPa) for 16 hours, followed by curing as in the first group.

2.2.3. Specimen testing method

The compressive strength tests were conducted using an electronic universal testing machine (DTU-900 model from Daekyung Tech and Tester). The axial strain was controlled at a

rate of 1 mm/min. The force values were measured at the sample failure point or 15 % axial strain. Each compressive strength test was conducted with 3 specimens. The compressive strength results were calculated according to ASTM D2166 -16.

Following the compressive strength tests, representative fragments were chosen from the fractured surface of the samples for a detailed examination of microstructure characteristics. This analysis was conducted by employing X-ray diffraction (XRD) on EMPYREAN using radiant Cu-K α ($\lambda = 1,5406$ Å). Fourier transform infrared spectroscopy (FTIR) technique with NICOLET 6700, wavenumber range for spectra was 400~4000 cm⁻¹, the resolution was 4 cm⁻¹. Scanning electron microscopy (SEM) observations were carried out using Hitachi S4800 instrument, specimens are coated with gold.

3. RESULTS AND DISCUSSION

3.1. Response surface design and statistical analysis

The average of the three-test obtained for each experimental combination was fitted to the general form of a quadratic polynomial model. Table 3 presents the results of the experimental design.

Table 3. The results of experimental design

Run	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
y ₁ (MPa)	7.3	0.89	9.43	1.02	8.87	1.23	7.9	1.04	5.18	5.71	3.18	4.35	5.18	5.76	5.08
y ₂ (MPa)	11.49	1.47	12.89	1.71	12.43	1.34	11.12	1.09	9.04	9.98	5.34	6.92	9.33	9.83	9.82

ANOVA performed response fit analysis, regression coefficient estimation, and model significance assessment with statistical software. The adequacy of the model was tested using the F value and the coefficient of determination (R^2). Variables with P-values less than 0.1 are considered valid and with P-values less than 0.05 are considered highly significant [28].

The strength of solidified materials cured under two conditions at 7d period was analyzed. The quadratic model was found to be fitted and could be adopted because it intuitively reflects the interaction between different variables. The regression model is shown in equations (1) and (2):

$$y_1 = 5.34 - 3.66x_1 + 0.50x_2 - 0.57x_3 - 0.50x_1x_2 + 0.20x_1x_3 + 0.16x_2x_3 - 0.26x_1^2 - 0.42x_2^2 - 0.32x_3^2$$
(1)

$$y_2 = 9.66 - 5.29x_1 + 0.52x_2 - 1.04x_3 - 0.29x_1x_2 + 0.27x_1x_3 + 0.16x_2x_3 - 2.05x_1^2 - 0.72x_2^2 - 1.12x_3^2$$
(2)

The statistics in Table 3 summarize the analysis of variance (ANOVA) of each model. The p-values of all models are less than 0.01, indicating that the regression effect is significant. The p-values for lack of fit are > 0.05 for all models, indicating abnormal errors in the fitted models and a good fit between the measured and predicted values. In addition, the high regression coefficients R^2 (> 0.98) show the compatibility of models, which means that there are more than 98% compatible experimental data with data in the model predictions. Moreover, the high

adjusted regression coefficients $adj-R^2$ confirmed that the model was highly feasible [29]. The contrast between $adj-R^2$ and pred- R^2 of all fitted models were less than 0.2, which indicates the consistency between $adj-R^2$ and pred- R^2 [30].

	y ₁ (MPa)	y ₂ (MPa)
Model		
\mathbb{R}^2	0.9920	0.9859
Adj-R ²	0.9775	0.9606
Pred-R ²	0.9035	0.7836
F-value	68.55	38.97
p-value	0.0001	0.0004
Lack of fit (p-value)	0.4030	0.0653

Table 4. ANOVA for response model of experimental results

Response surface plots can express regression models, including contour and surface plots. They also can directly visualize the combined effects of two factors on the response value, by keeping other factors at the middle level. Figures 1 and 2 show the response surfaces of solidified materials under room temperature and hydrothermal condition at 7d.



Figure 1. Surface plot for compressive strength of solidified materials under room temperature condition.



Figure 2. Surface plot for compressive strength of solidified materials under hydrothermal condition.

The effect of interaction between the DS and PC/FA ratios on compressive strength values can be shown as Figures 1(a) and 2(a). When the ratio of PC/FA remains constant, the

compressive strength increases approximately linearly as the DS ratio decreases, the curved surface is steep. At a constant DS ratio, the compressive strength increases with increasing PC/FA ratio, but it's not a significant change. This indicates that the DS ratio is the main factor influencing material compression strength. This is because, DS has poor geotechnical properties (low strength and stiffness), the high DS ratio means the low (PC+FA) ratio, that main components have hydraulic activity. A similar result with the interaction of DS ratio and water ratio as shown in Figures 1(b) and 2(b).

Figures 1(c) and 2(c) depict the impact of the PC/FA ratio and water ratio on compressive strength values, the surface is flat, showing the weak effect of two factors. Compressive strength values increase when the PC/FA ratio increases for the constant water ratio. If the PC/FA ratio is unchanged, the water ratio negatively correlates with compressive strength values.

3.2. Optimize the proportion of DS in the mixture and applicability

In accordance with the response surface fitting model and the desirability function, determine the DS ratio, PC/FA ratio, and water ratio to achieve the target compressive strength value at the maximum DS ratio and minimum PC/FA ratio. Two factors are considered equally important and less than the target value of compressive strength.

It is defined that the desirability function (d_i) of a single response range from 0 to 1, with 0 representing the unsatisfactory response value and 1 representing the response value at the desired level. The total desirability function (D) is the power of each response craving value multiplied exponentially. The goal of multi-objective optimization is to maximize the total desirability function D, the solution was solved by Design Expert software.

Target values are selected based on technical specifications for concrete blocks according to TCVN 6477:2016. This work uses a strength target of 7.5 MPa after 7d cured for two conditions. Table 5 and 6 summarize the optimization criteria and solution of two desirability functions, D_1 and D_2 .

<i>Table 5.</i> Optimization criteria for d	lifferent factors, responses,	, and solutions for the des	sirability function
D_1 , compressive streng	gth of materials cured roon	n temperature condition a	at 7d.

Factors	Goal	Lower limit	Upper limit	Solution
x_1 : DS / (DS+PC+FA) ratio	Maximize	0.3	0.9	0.420
x ₂ : PC/FA ratio	Minimize	2	4	2.609
x_3 : Water (W) / (DS+PC+FA) ratio	In range	0.4	0.5	0.400
y ₁ : compressive strength at 7d, room temperature curing (MPa)	Target 7.5	0.96	8.99	7.50

As can be seen in Table 5 and Table 6, it obtained that optimal ratio with desirability function value D_1 is 0.754 and D_2 is 0.932, indicating that the projected value is very reliable [29]. In addition, when specimens were cured under hydrothermal condition, a higher DS ratio and minimum PC/FA ratio were used for the same compressive strength target compared to cured room temperature condition. The main constituent of PC is clinker that fired at 1450 °C. This result indicates the positive effect of hydrothermal curing due to increasing DS content and high PC ratio substituted by FA. That means a significant decrease in the CO₂ emission of the mixture and an increase in dredged sediment treatment capacity in the same period.

Factors	Goal	Lower limit	Upper limit	Solution
x_1 : DS / (DS+PC+FA) ratio	Maximize	0.3	0.9	0.666
x ₂ : PC/FA ratio	Minimize	2	4	2.0
x ₃ : Water (W) / (DS+PC+FA) ratio	In range	0.4	0.5	0.424
y ₂ : compressive strength at 7d, hydrothermal curing (MPa)	Target 7.5	0.98	12.46	7.50

Table 6. Optimization criteria for different factors and responses and solutions for desirability function D₂, compressive strength of materials cured hydrothermal condition at 7d

An experiment performed using these proportions of components in mixtures D_1 and D_2 gave 7.21±0.11 (MPa) and 7.63±0.08 (MPa), respectively, room temperature and hydrothermal curing conditions, close to the target value. The model was proved to be validated at these points.

Table 7 shows the CO_2 emission of materials in the mixture, assuming that the CO_2 emission of dredged sediment is approximately zero and the proportion of components in mixtures D_1 and D_2 from the above result. Calculated CO_2 emissions in mixtures D_1 and D_2 are 0.127 and 0.066 kg CO_2 /kg, respectively; the same observation of low CO_2 emissions was reported by Bhairappanavar [31].

Table 7. CO_2	emissions and	proportion of	of components	in mixtures D ₁ and D	2
-		1 1	1	-	-

Materials	kgCO ₂ /kg	Mixture D_1 (%)	Mixture D_2 (%)
Dredged sediments	/	30.00	46.77
Portland cement	0.419 [32]	29.95	15.64
Fly ash	0.008 [33]	11.47	7.82
Water	0.001 [33]	28.58	29.77

3.3. Microstructure properties

To further investigate the mechanism of mechanical performance, samples were prepared from mixtures with DS ratios of 0.3 and 0.9, PC/FA ratios of 4, and water ratios of 0.45, respectively, with run order specimens 3 and 4, which were selected for characterization and analysis.

The XRD analysis results on solidified materials samples are presented in Figure 3(a). The occurrence of reaction from cement binder in the mixture under two curing conditions forms new crystalline phases such as ettringite (PDF # 41–1451), portlandite (PDF # 72–0156), katoite (PDF # 77–1713), tobermorite (PDF # 19–1364), which improve compressive strength performance of solidified materials. Ettringite, portlandite, and C-S-H gel are the main products of cement hydration. The C-S-H phases exhibit a tobermorite or genite structure at the nanoscale, as indicated by previous studies [34, 35]. Their structure is characterized by poor crystallinity and short-range properties, which have limited the detection in XRD patterns. Under the hydrothermal curing regime, there were some reactions that changed the minerals in the materials. Formation of tobermorite and katoite in 3 – hydrothermal sample through the reaction of portlandite and quartz [36], shown by the disappearance of portlandite diffraction peak and decreased diffraction intensity in quartz.



Figure 3. (a) XRD pattern and (b) FTIR spectra of solidified materials.

The FTIR spectra of solidified materials under curing conditions are shown in Figure 3(b), there is a similar trend in the FTIR spectra. Noteworthy, the FTIR band of sample 3 – Hydrothermal at ~1170 cm⁻¹ is characteristic of Al-tobermorite and is assigned to Si-O and/or Al-O stretching of Q_3 -silicate (aluminate) tetrahedra [37]. The asymmetric Si–O stretching mode at 970 cm⁻¹ indicates the formation of silicate chains containing Q_2 units [38]. It confirmed the presence of tobermorite under hydrothermal curing, which is consistent with the XRD result.



Figure 4. SEM images of (a) 3 – Room temperature sample and (b) 3 – Hydrothermal sample.

Figure 4 is the SEM image of the fracture surfaces of solidified materials under two curing conditions after 7d. These images confirmed the presence of the cementitious component and structural differences due to curing conditions. Figure 4(a) shows dredged sediments and fly ash particles embedded in the matrixes of gelatinous hydrates with large pores. In Figure 4(b), it can be observed that the sample has a denser structure with fewer pores, and the grain boundaries are more complex than in Figure 4(a). It reveals the presence of distinct, band-like crystals of tobermorite, approximately 1 μ m in size, forming interlocked honeycomb structure. This result points out the role of tobermorite in enhancing the physical performance of solidified materials by porosity filling with crystalline phase under hydrothermal condition [39].

4. CONCLUSIONS

In this study, the effects of three variables (DS/(DS+PC+FA) ratio, PC/FA ratio, and W/(DS+PC+FA) ratio) on the compressive strength of solidified materials were investigated

using the Response Surface Method according to Box-Behnken design. These models are very compatible, providing a method to design the mixing ratio. The component proportions to achieve a compressive strength of 7.5 MPa in 7 days, while maximizing the DS ratio and minimizing the PC/FA ratio, are as follows: for room temperature curing, 30.0 % DS, 29.95 % PC, 11.48 % FA, and 28.57 % water; for hydrothermal curing, 46.77 % DS, 15.64 % PC, 7.82 % FA, and 29.77 % water. Estimated CO₂ emissions for mixtures of room temperature and hydrothermal curing are 0.127 kg CO₂/kg and 0.066 kg CO₂/kg, respectively. Hence, hydrothermal curing presents a promising new treatment method, offering improved processing capacity and increased environmental friendliness.

Characterization of the microstructure of the solidified materials by XRD, FTIR, and SEM analysis showed that hydrothermal curing accelerated the formation of tobermorite, which contributed to the strength improvement by filling the pore spaces.

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Declaration of competing interest. The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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