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# NEOLIGNANS AND FLAVONOL GLYCOSIDES FROM THE LEAVES OF VIBURNUM LUTESCENS BLUME

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Abstract. Viburnum lutescens Blume, known as Vot vang nhat in Vietnamese, is a perennial shrub widely distributed South America, China and Southeast Asia. This plant has been used by folk medicine to treat various diseases, such as, cough, diarrhea, rheumatoid arthritis. V. lutescens has not been studied for its chemical composition and biological activity. In our study, we used different chromatographic techniques including column chromatography, HPLC, thin layer chromatography, ion exchange chromatography, and size exclusion chromatography to isolate the components. The chemical structures of the isolates were elucidated by using a combination of spectroscopic techniques such as 1D- and 2D-nuclear magnetic resonance spectra, electron spray ionization mass spectroscopy and by comparison with those reported in the literature. As the results, four neolignans including (7S,8S)-dihydrodehydrodiconiferyl alcohol 9-O- $\beta$ -D-glucopyranoside (1), (7*S*,8*S*)-dihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -D-(7*S*,8*R*)-5-methoxydihydrodehydrodiconiferyl alcohol 4-*O*-β-Dglucopyranoside (2),glucopyranoside (3) and (7S,8R)-5-methoxydihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -Dglucopyranoside and five flavonol glycosides including kaempferol-3-O- $\beta$ -D-(4), 3-O-rutinoside kaempferol kaempferol glucopyranoside (5), (6), 3-*O*-α-Lrhamnopyranosyl( $1 \rightarrow 6$ )- $\beta$ -D-galactopyranoside (7), kaempferol 3-O- $\beta$ -D-galactopyranoside (8), and quercetin 3-O- $\beta$ -D-glucopyranoside (9) were isolated from the methanol extract of Viburnum lutescens Blume leaves. Interestingly, compounds 1, 2, and 4 were reported from Viburnum genus for the first time. This study might be a good start to further investigation on the chemical as well as biological activity of this plant.

Keywords: Viburnum lutescens, Caprifoliaceae, neolignan, flavonolglycoside.

Classification numbers: 1.1.1, 1.1.6.

# **1. INTRODUCTION**

The genus Viburnum (Caprifoliaceae) consists of over 200 species (8 Viburnum species distribute in Viet Nam). Viburnum species are used in the folk medicines to treat cough,

diarrhea, rheumatoid arthritis, and tumefaction [1, 2]. Phytochemical investigations of *Viburnum* species indicated the presence of monoterpenoids, sesquiterpenoids, diterpenoids, triterpenoids, iridoids, flavonoids, and lignans, etc. [1]. *V. lutescens* has been used for the treatment of rheumatism, wounds, and obstructive inflammation of the veins [2]. However, chemical constituents and bioactivity of *V. lutescens* have not been studied. We report here the isolation and elucidation of four neolignans and five flavonol glycosides from the leaves of *V. lutescens*.

#### 2. MATERIALS AND METHODS

#### 2.1. Plant materials

The leaves of *Viburnum lutescens* Blume were collected in Vinh Phuc province, Vietnam on May 15<sup>th</sup>, 2019 and identified by Dr. Nguyen The Cuong, Institute of Ecology and Biological Resources, VAST. A voucher specimen was deposited at University of Science and Technology Hanoi, VAST.

### 2.2. General experimental procedures

Optical rotations were determined on a Jasco DIP-370 automatic polarimeter. The NMR spectra were recorded using a Bruker DRX 500 spectrometer (<sup>1</sup>H, 500 MHz; <sup>13</sup>C, 125 MHz). ESI-MS spectra were obtained using an Agilent 1100 LC-MSD Trap system. Column chromatography was performed using silica gel 60 (70 - 230 mesh or 230 - 400 mesh, Merck, Germany) or RP-18 resins (30 - 50  $\mu$ m, Fujisilisa Chemical Ltd., Japan). Pre-coated silica gel 60 F<sub>254</sub> (0.25 mm, Merck) and RP-18 F<sub>254</sub>S plates (0.25 mm, Merck) were used for thin-layer chromatography (TLC).

#### 2.3. Extraction and isolation

The dried powder of V. lutescens leaves (7.0 kg) was sonicated 3 times with hot MeOH (15 L each) to give a MeOH extract (VLL, 500 g) after removal of solvent under reduced pressure. VLL extract was suspended with water (4.0 L) then partitioned with dichloromethane and ethyl acetate to give dichloromethane (VLL1, 110.0 g), ethyl acetate (VLL2, 3.5 g), and water layer (VLL3). VLL3 was chromatographed on a Diaion HP-20 column eluting with water to remove sugar components, then with increasing concentration of methanol in water (25 % and 100 %) to obtain two fractions, VLL3A and VLL3B, respectively. VLL3B (32.0 g) was separated on a silica gel column eluting with gradient dichloromethane/methanol  $(1/0 \rightarrow 0/1, v/v)$  to give five fractions, VLL3B1 - VLL3B5. VLL3B3 fraction (0.8 g) was chromatographed on an RP-18 column eluting with methanol/water (1/1, v/v) to give four fractions, VLL3B3A – VLL3B3D. Compound 9 (16.1 mg) was obtained from VLL3B3A by a silica gel column eluting with ethyl acetate/methanol/water (8/1/0.1, v/v/v). VLL3B3C fraction was chromatographed on an HPLC system (J'sphere H-80 250 mm length  $\times$  20 mm ID column, eluting with 60 % acetonitrile in water, a flow rate of 3 mL/min) to yield compounds 2 (18.7 mg,  $t_R$  50.1 min) and 1 (9.4 mg,  $t_R$ 54.3 min). Compound 4 (3.6 mg, t<sub>R</sub> 51.5 min) was obtained from VLL3B3D fraction on HPLC system (J'sphere H-80 250 mm length  $\times$  20 mm ID column) eluting with 20 % ACN in water, a flow rate of 3 mL/min). VLL3B5 fraction was chromatographed on an RP-18 column eluting with methanol/water (1/3, v/v) to give four fractions, VLL3B5A - VLL3B5D. Compound 3 (12.3 mg, t<sub>R</sub> 56.1 min) was yielded from VLL3B2B on HPLC system (J'sphere H-80 250 mm length  $\times$  20 mm ID column, eluting with 20 % ACN in water, a flow rate of 3 mL/min).

VLL3B5 was chromatographed on an HPLC system (J'sphere H-80 250 mm  $\times$  20 mm column, eluting with 18 % ACN in water, a flow rate of 3 mL/min) to yield compounds 7 (16.6 mg, t<sub>R</sub> 40.8 min), **6**(33.9 mg, t<sub>R</sub> 46.1 min), **8** (3.1 mg, t<sub>R</sub> 53.4 min), and **5** (5.2 mg, t<sub>R</sub> 60.6 min).



Figure 1. Chemical structures of compounds 1-9.

(7*S*,8*S*)-*Dihydrodehydrodiconiferyl alcohol* 9-*O*- $\beta$ -*D*-*glucopyranoside* (1): Yellowish powder; ESI-MS: *m*/*z* 523.49 [M+H]<sup>+</sup>; C<sub>26</sub>H<sub>34</sub>O<sub>11</sub>;  $[\alpha]_D^{25}$ : -20.7 (*c* = 0.1, MeOH); <sup>1</sup>H- and <sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1.

(7S,8S)-Dihydrodehydrodiconiferyl alcohol 9'-O-β-D-glucopyranoside (2): Yellowishpowder; ESI-MS: m/z 523.41 [M+H]<sup>+</sup>; C<sub>26</sub>H<sub>34</sub>O<sub>11</sub>; [α]<sub>D</sub><sup>25</sup>: -18.3 (c = 0.1, MeOH);<sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 6.97 (d, J = 2.0 Hz, H-2), 6.78 (d, J = 8.0 Hz, H-5), 6.85 (dd, J = 2.0, 8.0 Hz, H-6), 5.51 (d, J = 6.5 Hz, H-7), 3.49 (m, H-8), 3.78 (dd, J = 5.0, 12.0 Hz, H<sub>a</sub>-9), 3.86 (dd, J = 2.5, 12.0 Hz, H<sub>b</sub>-9), 6.77 (s, H-2'), 6.78 (s, H-6'), 2.69 (t, J = 7.5 Hz, H-7'), 1.92 (m, H-8'), 3.56 (m, H<sub>a</sub>-9'), 3.95 (m, H<sub>a</sub>-9'), 4.27 (d, J = 7.5 Hz, H-1"), 3.23 (dd, J = 7.5, 9.0 Hz, H-2"), 3.39 (t, J = 9.0 Hz, H-3"), 3.37 (t, J = 9.0 Hz, H-4"), 3.28 (m, H-5"), 3.69 (dd, J = 5.5, 11.5 Hz, H<sub>a</sub>-6"), 3.88 (dd, J = 2.5, 11.5 Hz, H<sub>b</sub>-6"), 3.83 (s, 3-OMe), and 3.87 (s, 3'-OMe);<sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1.

(7*S*,8*R*)-5-Methoxydihydrodehydrodiconiferylalcohol 4-O-β-D-glucopyranoside (3): Yellowish powder; ESI-MS: m/z 553.37 [M+H]<sup>+</sup>; C<sub>27</sub>H<sub>36</sub>O<sub>12</sub>;  $[\alpha]_D^{25}$ :+26.1 (c = 0.1, MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 6.65 (s, H-2), 6.65 (s, H-6), 5.47 (d, J = 6.0 Hz, H-7), 3.35 (m, H-8), 3.66 (m, H<sub>a</sub>-9), 3.78 (m, H<sub>b</sub>-9), 6.62 (s, H-2'), 6.59 (s, H-6'), 2.54 (t, J = 6.5 Hz, H-7'), 1.73 (m, H-8'), 3.47 (t, J = 6.5Hz, H-9'), 4.77 (d, J = 7.5Hz, H-1''), 3.39 (dd, J = 7.5, 9.0 Hz, H-2''), 3.12 (m, H-3''), 3.30 (t, J = 9.0 Hz, H-4''), 3.31 (m, H-5''), 3.57 (m, H<sub>a</sub>-6''), 3.70 (m, H<sub>b</sub>-6''), 3.78 (s, 3-OMe), 3.78 (s, 5-OMe), and 3.73 (s, 3'-OMe); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1.

(7S,8R)-5-Methoxydihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -D-glucopyranoside (4): Yellowish powder; ESI-MS: m/z 575.23 [M+Na]<sup>+</sup>; C<sub>27</sub>H<sub>36</sub>O<sub>12</sub>,  $[\alpha]_D^{25}$ : +21.8 (c = 0.1, MeOH)<sup>1</sup>H- NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 6.97 (d, J = 2.0 Hz, H-2), 6.78 (d, J = 8.0 Hz, H-5), 6.85 (dd, J = 2.0, 8.0 Hz, H-6), 5.51 (d, J = 6.5 Hz, H-7), 3.49 (m, H-8), 3.78 (dd, J = 5.0, 12.0 Hz, H<sub>a</sub>-9), 3.86 (dd, J = 2.5, 12.0 Hz, H<sub>b</sub>-9), 6.77 (s, H-2'), 6.78 (s, H-6'), 2.69 (t, J = 7.5 Hz, H-7'), 1.92 (m, H-8'), 3.56 (m, H<sub>a</sub>-9'), 3.95 (m, H<sub>b</sub>-9'), 4.27 (d, J = 7.5 Hz, H-1"), 3.22 (dd, J = 7.5, 9.0 Hz, H-2"), 3.38 (t, J = 9.0 Hz, H-3"), 3.37 (t, J = 9.0 Hz, H-4"), 3.28 (m, H-5"), 3.69 (dd, J = 5.5, 11.5 Hz, H-6"), 3.87 (dd, J = 2.5, 11.5 Hz, H-6"), 3.88 (s, 3-OMe), 3.83 (s, 3'-OMe) and 3.88 (s,5-OMe); <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 1.

*Kaempferol 3-O-β-D-glucopyranoside* (5): Yellow powder;  $[\alpha]_D^{25}$ : -61.3 (c = 0.1, MeOH); ESI-MS m/z 448.42 [M+H]<sup>+</sup>; <sup>1</sup>H-and <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 2.

*Kaempferol 3-O-rutinoside* (6): Yellow powder; ESI-MS: m/z 595.19 [M+H]<sup>+</sup>;  $C_{27}H_{30}O_{15}$ ;  $[\alpha]_D^{25}$ : -17.8 (c = 0.1, MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{H}$ : 6.22 (s, H-6), 6.41 (s, H-8), 7.08 (d, J = 8.5 Hz, H-2', 6'), 6.91 (d, J = 8.5 Hz, H-3', 5'), 5.14 (d, J = 7.5 Hz, H-1''), 3.48 (dd, J = 7.5, 9.0 Hz, H-2''), 3.45 (t, J = 9.0 Hz, H-3''), 3.26 (t, J = 9.0 Hz, H-4''), 3.37 (m, H-5''), 3.83 (br d, J = 11.5 Hz,  $H_a$ -6''), 3.43 (dd, J = 4.5, 11.5 Hz,  $H_b$ -6''), 4.54 (brs, H-1'''), 3.66 (br d, J = 2.5 Hz, H-2'''), 3.49 (dd, J = 2.5, 9.0 Hz, H-3'''), 3.27 (t, J = 9.0 Hz, H-4'''), 3.43 (m, H-5'''), and 1.14 (d, J = 6.0 Hz, H-6'''); <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 2.

*Kaempferol* 3-O-α-L-rhamnopyranosyl-(1→6)-β-D-galactopyranoside (7): Yellow powder; ESI-MS: m/z 615.32 [M+Na]<sup>+</sup>; C<sub>28</sub>H<sub>32</sub>O<sub>14</sub>;  $[\alpha]_D^{25}$ : -22.7 (c = 0.1, MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 8.10 (d, J = 8.4Hz, H-2'), 6.91 (d, J = 8.4 Hz, H-3'), 6.91 (d, J = 8.4Hz, H-5'), 8.10 (d, J = 8.4 Hz, H-6'), 6.40 (br s, H-6), 6.21 (br s, H-8), 5.03 (d, J = 8.0 Hz, H-1"), 4.54 (br s, H-1"), and 1.16 (d, J = 6.4 Hz, H-6''); <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 2.

*Kaempferol* 3-*O*-β-*D*-galactopyranoside (8): Yellow powder; ESI-MS: m/z 471.64 [M+Na]<sup>+</sup>; C<sub>21</sub>H<sub>20</sub>O<sub>11</sub>;  $[\alpha]_D^{25}$ : +17.9 (c = 0.1, MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 6.17 (d, J = 2.0 Hz, H-6), 6.35 (d, J = 2.0 Hz, H-8), 6.90 (d, J = 9.0 Hz, H-2', H-5'), 8.10 (d, J = 9.0 Hz, H-3'), 6.90 (d, J = 9.0 Hz, H-5'), 8.10 (d, J = 9.0 Hz, H-6'), and 5.10 (d, J = 8.0 Hz, H-1"); <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 2.

*Quercetin 3-O-β-D-glucopyranoside (9)*: Yellow powder; ESI-MS: m/z 465.10 [M+H]<sup>+</sup>; C<sub>21</sub>H<sub>20</sub>O<sub>12</sub>;  $[\alpha]_D^{25}$ : -66.5 (c = 0.1, MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta_{\rm H}$ : 6.22 (d, J = 1.5 Hz, H-6), 6.41 (d, J = 1.5 Hz, H-8), 7.73 (d, J = 1.5 Hz, H-2'), 6.89 (d, J = 8.5 Hz, H-5'), 7.63 (dd, J = 8.5, 1.5 Hz, H-6'), 5.25 (d, J = 7.5 Hz, H-1"), 3.36 (dd, J = 9.0, 7.5 Hz, H-2"), 3.47 (t, J = 9.0 Hz, H-3"), 3.50 (t, J = 9.0 Hz, H-4"), 3.24 (m, H-5"), 3.58 (dd, J = 5.0, 12.0 Hz, H<sub>a</sub>-6"), and 3.73 (dd, J = 2.0, 12.0 Hz, H<sub>b</sub>-6"); <sup>13</sup>C-NMR (CD<sub>3</sub>OD) data: see Table 2.

## **3. RESULTS AND DISCUSSION**

Compound **1** was obtained as a yellowish powder. The <sup>1</sup>H-NMR spectrum of **1** showed signals corresponding to an ABX spin system at  $\delta_{\rm H}$  7.01 (1H, d, J = 2.0 Hz), 6.88 (1H, dd, J = 2.0, 8.0 Hz), 6.79 (1H, d, J = 8.0 Hz), two *meta* aromatic protons at  $\delta_{\rm H}$  6.76 (1H, br s) and 6.75 (1H, br s), an anomeric proton at  $\delta_{\rm H}$  4.37 (1H, d, J = 8.0 Hz), and two methoxy groups at  $\delta_{\rm H}$  3.83 and 3.88 (each 3H, s). The <sup>13</sup>C-NMR and HSQC spectra of **1** showed signals of twenty six carbons including six oxygenated carbons ( $\delta_{\rm C}$  104.6, 78.3, 78.1, 75.2, 71.7, and 62.8) which were assigned to a glucopyranosyl moiety, twelve aromatic carbons at  $\delta_{\rm C}$  149.0, 147.5, 147.5, 145.2, 134.7, 136.9, 129.7, 119.7, 118.2, 116.1, 114.2, and 110.7 assigned to two benzene rings, one methine at  $\delta_{\rm C}$  55.4, one oxymethine at  $\delta_{\rm C}$  88.9, two oxymethylenes at  $\delta_{\rm C}$  72.5 and 62.2, and two methoxycarbons at  $\delta_{\rm C}$  56.8 and 56.4. Analysis of <sup>1</sup>H-, <sup>13</sup>C-NMR, and HSQC spectra of **1** 

indicated the structure of **1** as a dihydrodehydrodiconiferyl alcohol glucoside. Comparing <sup>13</sup>C-NMR chemical shifts of two oxygenated methylene groups at C-9 ( $\delta_{\rm C}$  72.5) and C-9'( $\delta_{\rm C}$  62.2) of **1** with those of (7*S*,8*S*)-dihydrodehydrodiconiferyl alcohol 9-*O*- $\beta$ -D-glucopyranoside [ $\delta_{\rm C}$  72.4 (C-9)/ $\delta_{\rm C}$  62.3 (C-9')] and (7*S*,8*S*)-dihydrodehydrodiconiferyl alcohol 9'-*O*- $\beta$ -D-glucopyranoside [ $\delta_{\rm C}$  65.1 (C-9)/ $\delta_{\rm C}$  70.0 (C-9')] confirmed the position of glucopyranosyl unit at C-9 [3]. Consequently, compound **1** was determined as (7*S*,8*S*)-dihydrodehydrodiconiferyl alcohol 9-*O*- $\beta$ -D-glucopyranoside [ $\beta_{\rm C}$  62.3 (C-9')] accompound **1** was determined as (7*S*,8*S*)-dihydrodehydrodiconiferyl alcohol 9-*O*- $\beta$ -D-glucopyranoside [3], a compound was reported from *Viburnum* genus for the first time.

	1			2		3		4	
С	${\delta_{\mathrm{C}}}^{\$,\mathrm{a}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	$\delta_{\rm H}^{\rm a,  c}(mult., J \text{ in Hz})$	${\delta_{\mathrm{C}}}^{^{\#,\mathrm{a}}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	${\delta_{\mathrm{C}}}^{*,\mathrm{a}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	${\delta_{\mathrm{C}}}^{^{@,\mathrm{a}}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$
1	134.8	134.7	-	134.9	134.8	140.3	140.3	134.1	134.1
2	110.9	110.7	7.01 (d, 2.0)	110.7	110.6	104.4	104.4	104.2	104.2
3	149.0	149.0	-	149.1	149.1	154.4	154.4	149.4	149.4
4	147.5	147.5	-	147.5	147.5	135.6	135.6	137.0	136.9
5	116.1	116.1	6.79 (d, 8.0)	116.2	116.2	154.4	154.4	149.4	149.4
6	119.8	119.7	6.88 (dd, 2.0, 8.0)	119.8	119.7	104.4	104.4	104.2	104.2
7	89.0	88.9	5.61 (d, 6.5)	89.0	89.0	88.5	88.5	89.1	89.1
8	55.3	55.4	3.49 (m)	55.4	55.4	55.8	55.8	55.6	55.6
9	72.4	72.5	3.80 (dd, 5.0, 12.0)	65.1	65.0	65.1	65.1	65.0	65.0
			4.23 (dd, 2.5, 12.0)						
1'	137.0	136.9	-	136.9	136.9	137.2	137.2	136.4	137.0
2'	114.4	114.2	6.75 (s)	114.4	114.3	114.2	114.3	114.3	114.4
3'	145.3	145.2	-	145.2	145.2	145.3	145.3	145.2	145.2
4'	147.5	147.5	-	147.5	147.5	147.4	147.5	147.5	147.5
5'	129.7	129.7	-	130.0	130.0	129.4	129.5	129.8	129.8
6'	118.3	118.2	6.76 (s)	118.1	118.1	117.9	118.0	118.0	118.1
7'	35.8	35.8	1.83 (m)	32.9	32.9	32.9	32.9	32.9	32.9
8'	32.9	32.9	2.64 (t, 7.5)	32.9	32.9	35.8	35.8	32.9	32.9
9'	62.3	62.2	3.58 (t, 6.5)	70.0	70.0	62.2	62.2	69.9	70.0
Glc									
1"	104.6	104.6	4.37 (d, 7.5)	104.5	104.5	105.2	105.2	104.5	104.5
2"	75.2	75.2	3.25 (dd, 7.5, 9.0)	75.2	75.2	75.7	75.7	75.2	75.2
3″	78.3	78.3	3.38 (t, 9.0)	78.2	78.1	78.3	78.3	77.9	78.2
4''	71.7	71.7	3.31 (t, 9.0)	71.8	71.7	71.3	71.3	71.7	71.7
5″	78.0	78.1	3.30 (m)	77.9	77.9	77.8	77.8	77.9	77.9
6″	62.9	62.8	3.68 (dd, 5.5, 11.5)	62.8	62.8	62.5	62.6	62.8	62.8
			3.88 (dd, 2.5, 11.5)						
3-OMe	56.5	56.4	3.83 (s)	56.4	56.8	57.0	57.0	56.8	56.8
5-OMe						57.0	57.0	56.8	56.8
3'-OMe	56.8	56.8	3.88 (s)			56.9	56.9		

*Table 1.* NMR spectral data for compounds 1 - 4 and reference compounds.

<sup>*a*</sup>)*measured in*  $CD_3OD$ , <sup>*b*</sup>)500MHz, <sup>*c*</sup>)125MHz,  $\delta_c^{\$}$  of (7S,8S)-dihydrodehydrodiconiferyl alcohol 9-O- $\beta$ -Dglucopyranoside [3],  $\delta_c^{\#}$  of (7S,8S)-dihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -D-glucopyranoside [3],  $\delta_c^{\$}$  of (7S,8R)-5-methoxydihydrodehydrodiconiferyl alcohol 4-O- $\beta$ -D-glucopyranoside [4],  $\delta_c^{@}$  of (7S,8R)-5methoxydihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -D-glucopyranoside [5].

Compound 2 was obtained as a yellowish powder. The <sup>13</sup>C-NMR data of 2 were found to be very similar to those of 1 except for two oxygenated methylenes [2: C-9 ( $\delta_{\rm C}$  65.0) and (C-9' ( $\delta_{\rm C}$  70.0); 1: C-9 ( $\delta_{\rm C}$  72.5) and C-9' ( $\delta_{\rm C}$  62.2)] (Table 1). This suggested the possibility of a

glucopyranosyl location at C-9'. The NMR data of **2** were matched with those of (7S,8S)dihydrodehydrodiconiferyl alcohol 9'-*O*- $\beta$ -D-glucopyranoside [3]. Thus, the structure of **2** was identified. This compound was also reported from *Viburnum* genus for the first time.

С			5	(	6		7		8		9	
	${\delta_{\mathrm{C}}}^{*,\mathrm{a}}$	$\delta_{\mathrm{C}}{}^{\mathrm{a,b}}$	$\delta_{\rm H}{}^{\rm a,c}(mult., J \text{ in Hz})$	${\delta_{\mathrm{C}}}^{^{@,\mathrm{a}}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	${\delta_{\mathrm{C}}}^{\mathrm{\$,a}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	${\delta_{\mathrm{C}}}^{\&,\mathrm{a}}$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	${\delta_{\mathrm{C}}}^{^{\#\!$	${\delta_{\mathrm{C}}}^{\mathrm{a,b}}$	
2	158.5	158.5	-	161.5	161.5	158.7	159.3	158.0	157.3	158.4	159.0	
3	135.5	135.5	-	135.5	135.5	135.7	135.7	134.6	134.1	135.6	135.6	
4	179.5	179.5	-	179.4	179.4	179.4	179.6	179.8	179.0	179.5	179.5	
5	163.1	163.1	-	163.0	163.0	162.9	163.0	162.1	160.2	163.1	163.1	
6	99.9	100.0	6.23 (brs)	100.0	100.1	-	100.0	98.8	98.8	99.9	99.9	
7	166.0	166.1	-	166.0	166.2	166.0	166.2	165.0	161.0	166.0	166.2	
8	94.7	94.8	6.42 (brs)	94.9	95.0	95.2	94.9	93.5	93.5	94.7	94.7	
9	159.1	159.1	-	158.6	158.6	159.2	158.5	157.5	155.4	158.5	158.5	
10	105.7	105.7	-	105.7	105.6	105.2	105.5	104.6	103.9	105.7	105.7	
1'	122.8	122.8	-	122.8	122.8	122.7	122.7	121.5	121.4	123.0	123.1	
2'	132.3	132.3	8.07 (d, 8.5)	132.4	132.4	132.4	132.5	131.1	130.9	116.0	116.0	
3'	116.1	116.1	6.91 (d, 8.5)	116.1	116.1	116.1	116.1	114.9	114.7	145.9	145.9	
4'	161.6	161.6	-	159.4	159.4	161.6	161.6	160.6	160.2	149.8	149.9	
5'	116.1	116.1	6.91 (d, 8.5)	116.1	116.1	116.1	116.1	114.9	114.7	117.6	117.6	
6'	132.3	132.3	8.07 (d, 8.5)	159.4	132.4	132.4	132.5	131.1	130.9	123.2	123.2	
3- <i>0-</i> Glc/Gal	Glc	Glc		Glc	Glc	Gal	Gal	Gal	Gal	Glc	Glc	
1″	104.1	104.2	5.26 (d, 7.0)	104.6	104.6	105.7	105.5	103.7	103.9	104.3	104.4	
2"	75.7	75.8	3.46 (dd, 7.0, 9.0)	75.8	75.8	73.0	73.0	71.7	71.7	75.7	75.7	
3‴	78.1	78.1	3.43 (dd, 9.0, 9.0)	78.2	78.2	75.1	75.0	73.7	73.7	78.1	78.1	
4″	71.4	71.4	3.35 (m)	71.4	71.5	70.2	70.1	68.7	68.6	71.2	71.2	
5″	78.4	78.4	3.23 (m)	77.3	77.2	75.4	75.3	75.8	75.7	78.4	78.4	
6″	62.6	62.7	3.71 (br d, 12.0) 3.55 (dd, 5.0, 12,0)	68.6	68.6	67.4	67.4	60.6	60.6	62.5	62.6	
6"-O-Rha												
1'''				102.4	102.4	101.9	101.9					
2'''				72.1	72.1	72.1	72.1					
- 3'''				72.3	72.3	72.3	72.3					
4'''				73.9	73.9	73.9	73.9					
5'''				69.7	69.7	69.7	69.7					
6'''				17.9	17.9	17.9	18.0					

Table 2. NMR spectral data for compounds 5-9 and reference compounds.

<sup>*a*</sup>)measured in CD<sub>3</sub>OD, <sup>*b*</sup>)500MHz, <sup>*c*</sup>)125MHz; \* $\delta_C$  of kaempferol-3-O- $\beta$ -D-glucopyranoside [6], <sup>*®*</sup> $\delta_C$  of kaempferol 3-Orutinoside [6], <sup>*§*</sup> $\delta_C$  of kaempferol 3-O-*a*-L-rhamnopyranosyl-(1 $\rightarrow$ 6)- $\beta$ -D-galactopyranoside [19], <sup>*®*</sup> $\delta_C$  of kaempferol 3-O- $\beta$ -D-galactopyranoside [11], <sup>*#*</sup> $\delta_C$  of quercetin 3-O- $\beta$ -D-glucopyranoside [12].

Compound **5** was obtained as a yellow amorphous powder. The <sup>1</sup>H-NMR spectrum of **5** suggested the presence of a flavonol glycoside with the appearance of two *meta* aromatic protons at  $\delta_{\rm H}$  6.23 (br s) and 6.42 (br s) (ring A) and two *para* aromatic protons at  $\delta_{\rm H}$  6.91 and 8.07 (each 2H, d, J = 8.5 Hz). The <sup>13</sup>C-NMR and DEPT spectra of **5** showed signals of 15 carbons of a kaempferol moiety at  $\delta_{\rm C}$  179.5, 166.1, 163.1, 161.6, 159.1, 158.5, 135.5, 132.3×2, 122.8,

116.1×2, 105.7, 100.0, and 94.8. The sugar moiety was determined as glucopyranoside based on the appearance of one anomeric proton at  $\delta_{\rm H}$  5.26 (d, J = 7.0 Hz) along with six carbons at  $\delta_{\rm C}$ 104.2, 78.4, 78.1, 75.8, 71.4, and 62.7. In addition, ESI-MS of **5** showed an anion peak at m/z449.42 [M+H]<sup>+</sup> corresponding to the molecular formula of C<sub>21</sub>H<sub>20</sub>O<sub>11</sub>. Based on above evidence and comparing NMR data of **5** with those reported in the literature (Table 2), the structure of **5** was identified as kaempferol 3-*O*- $\beta$ -D-glucopyranoside [6]. This compound was reported from *Wilkstroemiaindica* [7] and *Clitoriaternatea* [6].

Comparing NMR data of **6** with those of **5** indicated that the structure of **6** was similar to that of **5**, except for the addition of a rhamnopyranosyl unit. This sugar unit was identified based on the chemical shifts at  $\delta_{\rm C}$  102.4 (C-1<sup>'''</sup>), 72.1 (C-2<sup>'''</sup>), 72.3 (C-3<sup>'''</sup>), 73.9 (C-4<sup>'''</sup>), 69.7 (C-5<sup>'''</sup>), 17.9 (C-6<sup>'''</sup>) and multiplicity of ananomeric proton [ $\delta_{\rm H}$  4.54 (br s)] and one secondary methyl group [ $\delta_{\rm H}$  1.14 (d, J = 6.0 Hz)] in the <sup>1</sup>H-NMR spectrum. Thus, **6** was defined as kaempferol 3-*O*-rutinoside [6]. This compound was reported from *V. plicatum* Thunb. var. *tomentosum* [8] and found to exhibit antioxidant, antibacterial, and cytotoxic activities [9].

The remaining compounds were elucidated as (7S.8R)-5methoxydihydrodehydrodiconiferyl alcohol  $4-O-\beta$ -D-glucopyranoside (3) [4], (7S, 8R)-5methoxydihydrodehydrodiconiferyl alcohol 9'-O- $\beta$ -D-glucopyranoside (4) [5], kaempferol 3-O- $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 6)- $\beta$ -D-galactopyranoside (7) [10], kaempferol-3-*O*-β-Dgalactopyranoside (8) [11], quercetin 3- $O-\beta$ -D-glucopyranoside (9) [12] by spectroscopic methods and a comparison with those reported in the literature. Among them, 4 was reported from Viburnum genus for the first time: 3 was also reported from V. awabuki [13] and V. plicatum var. plicatum f. plicatum [14], 7 from V. punctatum [15], 8 from V. tinus [16], and 9 from V. dilatatum [17] and V. erosum [18].

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#### REFERENCES

- 1. Wang X. Y., Shi H. M., Li X. B. Chemical constituents of plants from the genus *Viburnum*, Chemistry & Biodiversity **7** (2010) 567-593.
- Chi V. V. Vietnamese dictionary of medical plants, Medical Publisher, Ha Noi, Vol. 1, 2012.
- 3. Lee S., Song I. H., Lee J. H., Yang W. Y., Oh K. B., Shin J. Sortase A inhibitory metabolites from the roots of *Pulsatilla koreana*, Bioorganic & Medicinal Chemistry Letters **24** (2014) 44-48.
- 4. Kuang H. X., Xia Y. G., Yang B. Y., Wang Q. H., Lü S. W. Lignan constituents from *Chloranthus japonicus* Sieb, Archives of Pharmacal Research **32** (2009) 329-334.
- 5. Shu J., Liang F., Zhu G., Liu X., Yu J., Huang H. Lignan glycosides from the rhizomes of *Smilax trinervula* and their biological activities, Phytochemistry Letters **20** (2017) 1-8.
- 6. Kazuma K., Noda N., Suzuki M. Malonylated flavonol glycosides from the petals of *Clitoria ternatea*, Phytochemistry **62** (2003) 229-237.

- Lee K. H., Tagahara K., Suzuki H., Wu R. Y., Haruna M., Hall I. H., Huang H. C., Ito K., Iida T., Lai J. S. - Antitumor agents. 49. Tricin, kaempferol-3-*O*-β-D-glucopyranoside and (+)-nortrachelogenin, antileukemic principles from *Wikstroemia indica*, Journal of Natural Products 44 (1981) 530-535.
- 8. Machida K., Sagawa H., Onoguchi R., Kikuchi M. Three new glycosides from *Viburnum plicatum* THUNB. var. *tomentosum* MIQ, Helvetica Chimica Acta **93** (2010) 290-297.
- Calderón-Montaño J. M., Burgos-Morón E., Pérez-Guerrero C., López-Lázaro M. A review on the dietary flavonoid kaempferol, Mini Reviews in Medicinal Chemistry 11 (2011) 298-344.
- 10. Hasan A., Ahmed I., Jay M., Voirin B. Flavonoid glycosides and an anthraquinone from *Rumex chalepensis*, Phytochemistry **39** (1995) 1211-1213.
- 11. Scharbert S., Holzmann N., Hofmann T. Identification of the astringent taste compounds in black tea infusions by combining instrumental analysis and human bioresponse, Journal of Agricultural and Food Chemistry **52** (2004) 3498-3508.
- 12. Park S. Y., Kim J., Lee S. Y., Bae K. H., Kang S. S. Chemical constituents of *Lathyrus davidii*, Natural Product Sciences **14** (2008) 281-288.
- 13. Matsuda N., Sato H., Yaoita Y., Kikuchi M. Isolation and absolute structures of the neolignan glycosides with the enantimetric aglycones from the leaves of *Viburnum awabuki* K. KOCH, Chemical & Pharmaceutical Bulletin **44** (1996) 1122-1123.
- 14. Katagiri S., Watanabe Y., Yaoita Y., Kikuchi M., Machida K. Two new phenolic glycosides from *Viburnum plicatum* var. *plicatum* f. *plicatum*, Natural Product Communications **6** (2011) 1901-1904.
- 15. Gao Y., Yang W. Q., Yang K., Jiang X. J., Li X. M., Wang F. New iridoid glycosides from *Viburnum punctatum*, Phytochemistry Letters **28** (2018) 145-148.
- Mohamed M. A., Marzouk M. S., Moharram F. A., El-Sayed M. M., Baiuomy A. R. -Phytochemical constituents and hepatoprotective activity of *Viburnum tinus*, Phytochemistry 66 (2005) 2780-2786.
- 17. Lu D., Yao S. Phenolic Glycoside from the Roots of *Viburnum dilatatum*, Natural Product Communications **4** (2009) 945-946.
- 18. Wu B., Wu S., Qu H., Cheng Y. Secondary metabolites from *Viburnum erosum*, Biochemical Systematics and Ecology **36** (2008) 817-819.
- Thinh N. S., Thu N. T. B., Khoi N. M., Ngoc T. M., Khoi N. M., Tai B. H., Kiem P. V., Nhiem N. X. - Flavonol glycosides from the leaves of *Fissitigma pallens* (Fin. & Gagn.) Merr, Vietnam Journal of Science and Technology 58 (2020) 526-532.